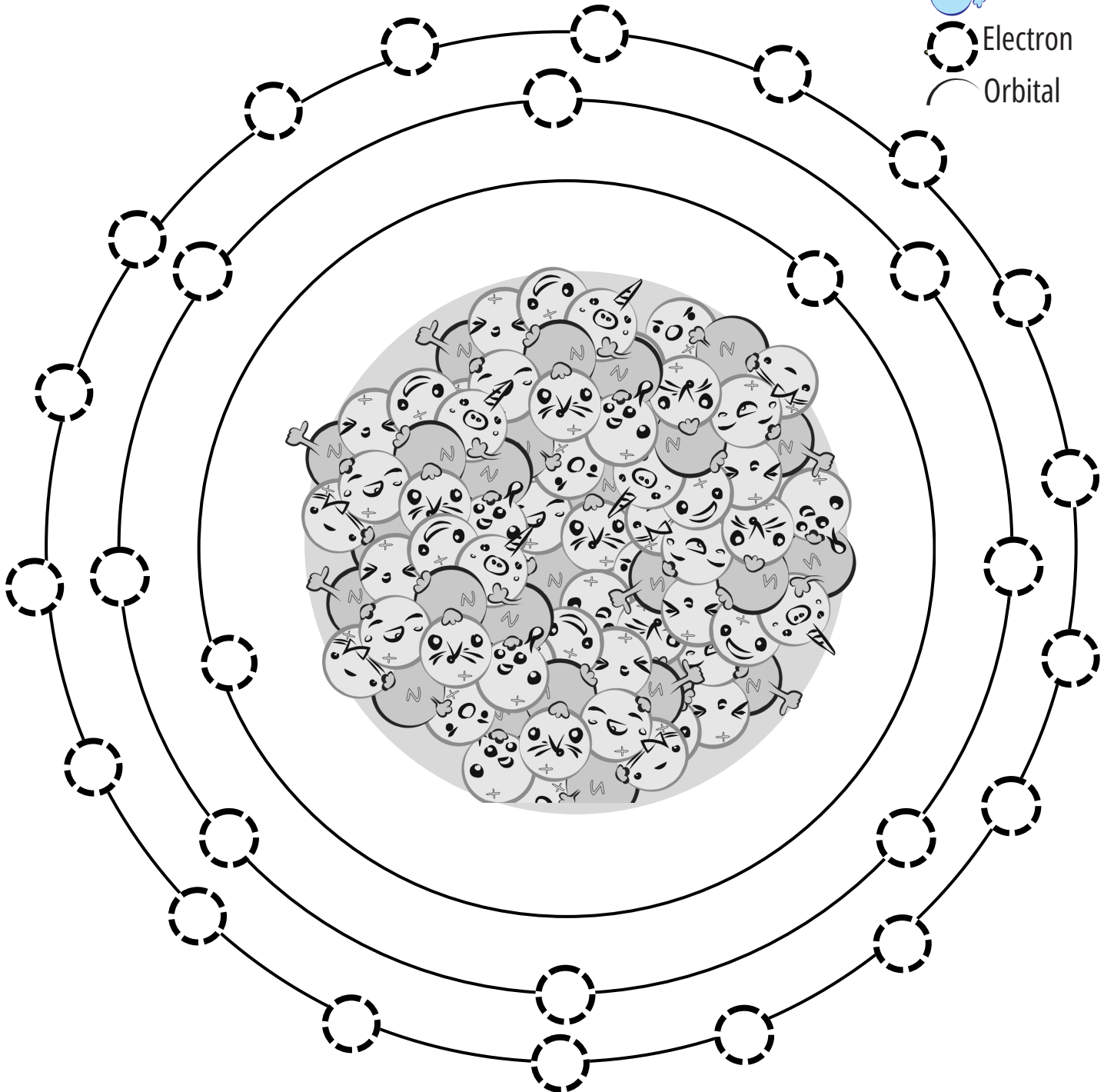
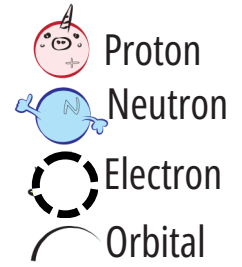


Model an Atom



Atoms make up everything we see in the world. The parts that make up an atom are called protons, neutrons and electrons. Protons and Neutrons are in the center. Electrons are zooming around them in different orbitals. Each orbital can hold a different number of electrons. Use a periodic table to choose elements with atomic numbers 1-18 to model your atom after. The atomic number for each element is the number of protons neutrons and electrons in each atom. For example, if the atomic number of Carbon is 6. There will be 6 protons, 6 electrons, and 6 neutrons.